





# Introduction to BMS

Course 3 | Module 1

















# Agenda

- 1. Cell Fundamentals & Operating Principles
- 2. Battery Basics
- 3. What is a BMS and its components
- 4. BMS Topologies







# Agenda

# 1. Cell Fundamentals & Operating Principles

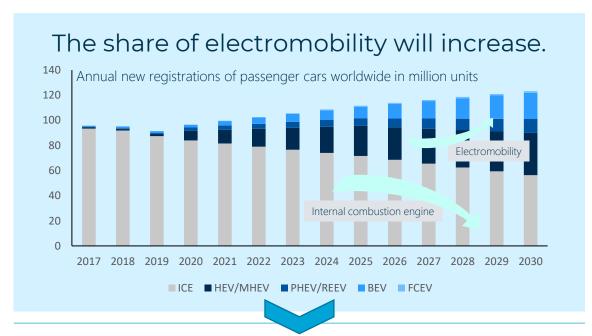
- 2. Battery Basics
- 3. What is a BMS and its components
- 4. BMS Topologies



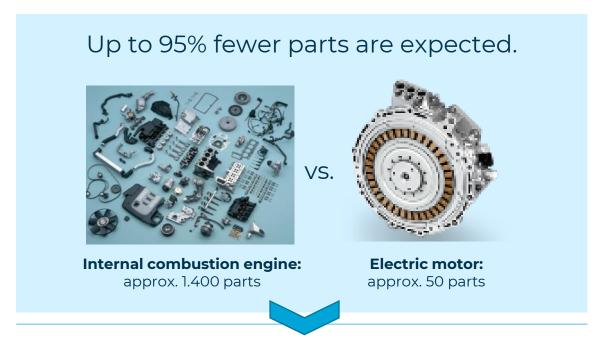




## We are on the verge of change



In the year 2030, 50% electric mobility is predicted.



In total, over 2000 parts will be eliminated from the system.

How can this change be addressed in your company?







### Basic Structure Of An Electric Vehicle



Source: Volkswagen

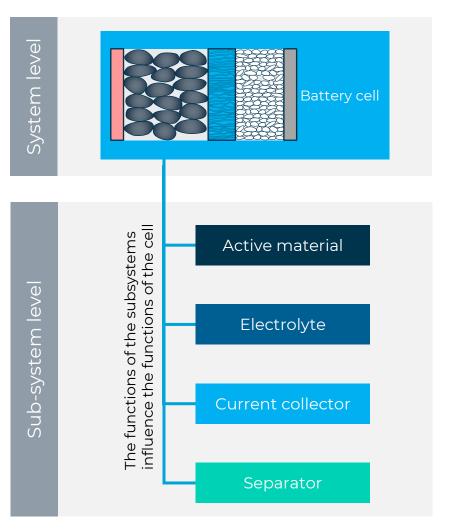
- 1 Charging connection
- 2 High-voltage (lithium-ion) battery
- 3 Power electronics
- 4 Electric motor
- 5 Battery management system
- 6 Engine compartment (drive & auxiliary units)
- 7 ABS & ESP
- 8 Recuperative braking system
- 9 High-voltage cables
- Low-voltage battery (12 V, DC converter)







## Any battery cell consists of four components



Core function of a battery cell

+ Reversible storage of electrical energy in chemical form

# Any lithium-ion cell, regardless of its cell type, consists of four components:

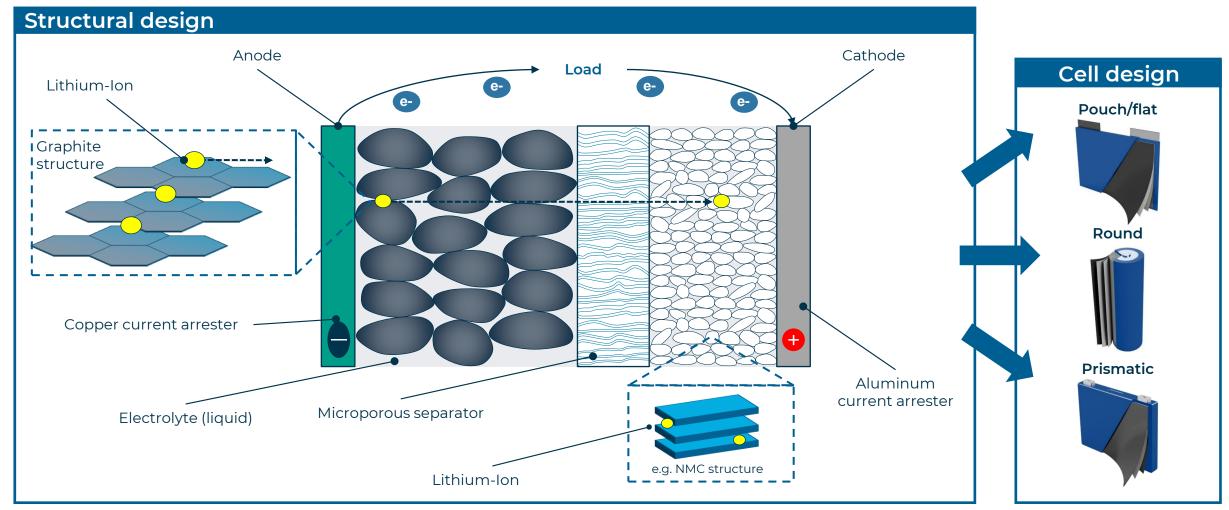
Active material	Electrolyte	Current collector	Separator
+ Task: Storage	+ Task: Transport	+ Task: Transport	+ Task: Electrical isolation
+ Place of chemical reaction	+ Enhancing ionic conductivity	+ Enhancing the electronic	+ Allows exchange of ions, but not electrons
+ Cathode: NMC	<ul> <li>Electrical insulator</li> </ul>	conductivity	electrons
+ Anode: Graphite	+ Consists of lithium	+ Anode side: Copper	+ Polyethylene (PE)
·	salt (e.g. LiPF <sub>6</sub> ) and organic solvent (e.g. DMC)	+ Cathode side: Aluminum	+ Polypropylene (PP)
			+ Thicknesses:
		+ Thicknesses: 8-20 µm	15-30 µm







# Structure Of A Lithium-ion Cell Section through the cell

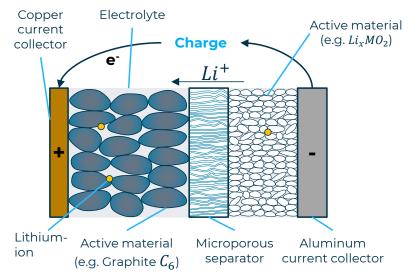


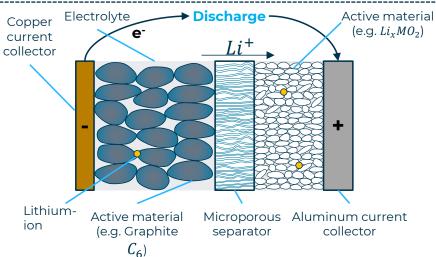






## Operating Principle Of A Lithium-ion Cell





#### **Operating principle**

- + Energy storage through **storage and removal of lithium ions** in the molecular lattices of the anode or cathode (intercalation / deintercalation)
- + The electrode at which the oxidation dominates during discharge is named the **anode** (negative pole), and the other, where the reduction dominates, is the **cathode** (positive pole). This nomenclature is valid only for the discharging reaction; for the charging reaction the names are reversed.
- + Simplified electrochemical reactions (e.g. Discharge):
  - + Negative electrode (during discharge "Anode")

$$Li_xC_6 \leftrightarrow xLi^+ + xe^- + C_6$$

+ Positive electrode (during discharge "Cathode")

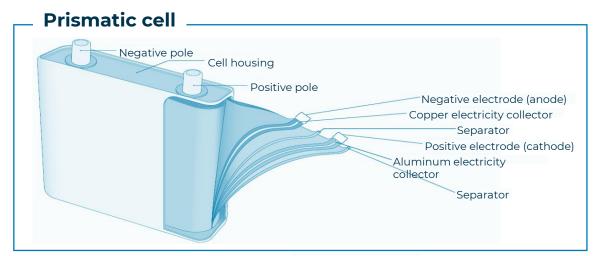
$$xLi^+ + xe^- + Li_{1-x}MO_2 \leftrightarrow Li_xMO_2$$
  
( $Li_xMO_2$ ,  $M$  e.g.  $Co$ ,  $Mn$ ,  $Ni$ )

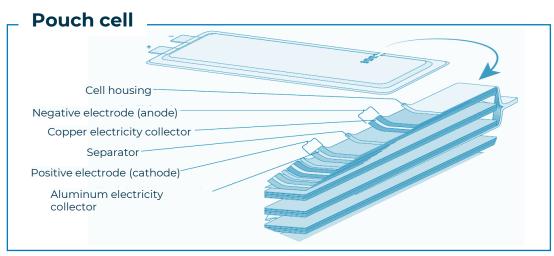


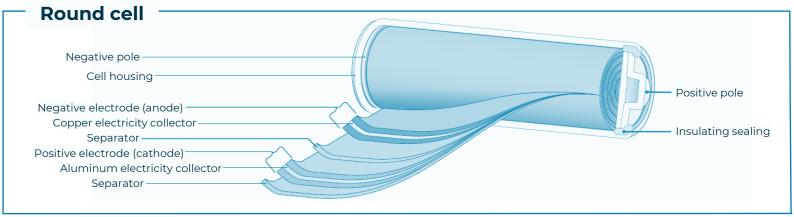




# Construction Of A Lithium-ion Battery Cell+ Same structure in all cell formats due to principle













# Advantages And Disadvantages Of Used Cell Formats

	Pouch cell	Round cell	Prismatic cell
Energy density	High at module level	Low at module level	Medium at module level
Life	Unconfirmed concerns	High	Good
Case	Aluminum foil	Metal	Plastic / Metal
Dimensions	<ul><li>+ Versatile design</li><li>+ Efficient use of space</li><li>+ High packing density</li></ul>	<ul><li>+ Lack of versatility</li><li>+ Inefficient use of space</li><li>+ Low packing density</li></ul>	<ul><li>+ Versatile design</li><li>+ Efficient use of installations</li><li>+ High packing density</li></ul>
Sturdiness	<ul><li>+ Low long-term tightness</li><li>+ Unstable housing</li><li>+ Swelling under pressure</li><li>+ Superstructure</li><li>+ Sealing very complex</li></ul>	<ul> <li>+ High tightness</li> <li>+ High rigidity</li> <li>+ Mechanical robust</li> <li>+ Resists certain internal pressure without deformation</li> </ul>	<ul> <li>+ High tightness</li> <li>+ Higher strength than pouch cell</li> <li>+ Lower mechanical stability than round cell</li> </ul>
Thermoregulation	<ul><li>+ Good surface-to-volume ratio</li><li>+ Efficient temperature control</li></ul>	+ High-energy cells: low heat dissipation	<ul> <li>Surface-to-volume ratio better than round cell, worse than pouch cell</li> </ul>
Special features of the production process	+ Difficult stacking	<ul><li>+ Extensive experience in the production process</li><li>+ Low manufacturing costs</li></ul>	<ul> <li>+ Modules without additional housing possible</li> <li>+ Pressure on cells necessary for joining</li> <li>+ Cell contacting only 1 step</li> </ul>
<b>Example</b> Image sources: BMW, Tesla, Nissan	Nissan Leaf	Tesla Model S	BMW i3

All three cell formats are used in various electric vehicles. With increasing maturity, a trend towards pouch cells is discernible.

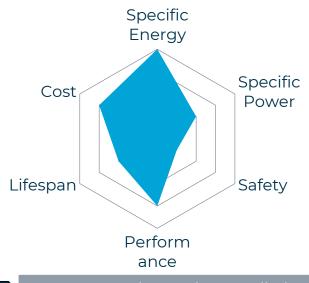






# Lithium-ion Chemistries, Applications, Benefits And Disadvantages

# Lithium ion Cobalt Oxide (LCO)



**Applications** 

Consumer Electronics as cell phones, laptops, and digital cameras

Benefits

High energy density resulting in smaller batteries

Disadvantages

Affected by high temperatures, short lifespan, typically last 500-1000 cycles

#### Lithium ion Manganese Oxide (LMO)

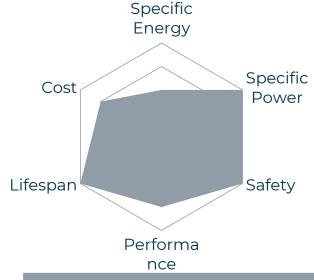


Power tools, medical devices, electric vehicles, and consumer electronics

Fast charging, high specific power, thermal stability, safety, and flexibility

Short lifespan of 300–700 charge cycles.

# Lithium iron phosphate battery (LiFePO<sub>4</sub>)



Energy Storage Systems, EVs, Marine and Backup Power Systems.

Long Lifespan of 2000-4000 cycles, and less prone to overheating

Lower energy density results in a heavier batteries

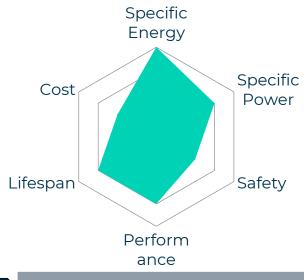






# Lithium-ion Chemistries, Applications, Benefits And Disadvantages

# Lithium Nickel Cobalt Aluminum Oxide (NCA)



Applications

EVs, Energy Storage Systems, industrial applications

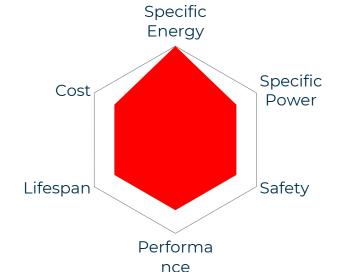
Benefits

High energy density resulting in smaller batteries and good lifespan

Disadvantages

Performance drops in extreme cold or hot conditions, high cost

# Lithium Nickel Manganese Cobalt Oxide (NMC)

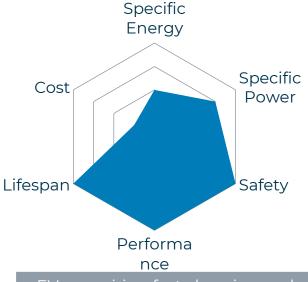


EVs, Energy Storage Systems,
Portable electronics, Industrial
equipment

High energy density, ratios of chemistries can be adjusted

Susceptible to capacity loss under high-stress conditions.

# Lithium Titanate Oxide (LTO)



EVs, requiring fast charging and durability, military and aerospace

Lifespan of 7000-20000 cycles, Wide Temperature range, Discharge Rates

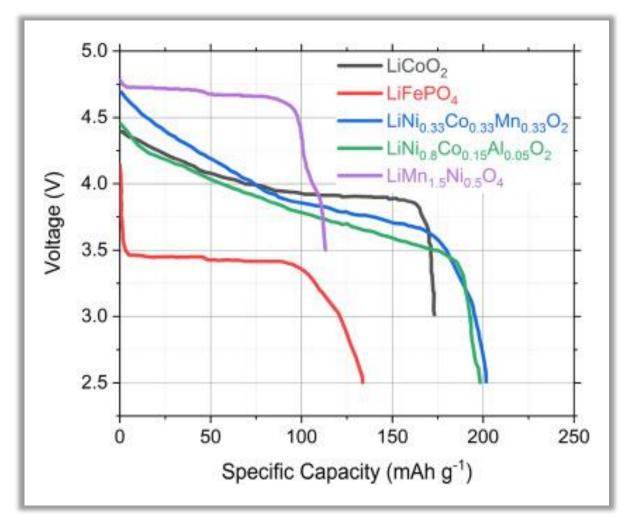
Higher cost and low energy density







# The discharge curve varies based on the cell's chemistry









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I. Cell Fundamentals & Operating Principles

# 2. Battery Basics

- 3. What is a BMS and its components
- 4. BMS Topologies







# The cells are assembled in several manufacturing steps, first into a module and then into a pack.

1 Battery cell

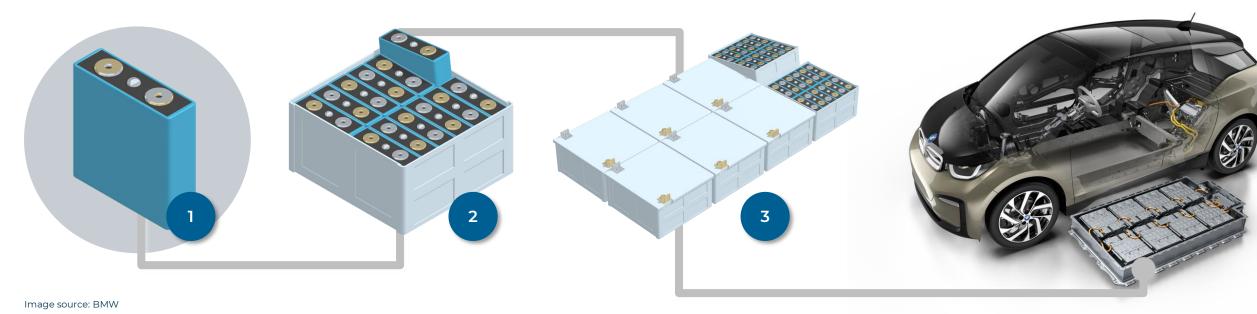
**Battery cells** are stacked (prismatic, pouch) or fixed in a **tray** (round).

2 Battery module

The cells are mounted and connected in the Battery module.

Battery pack

The modules are then assembled in the battery pack together with the peripheral system components.

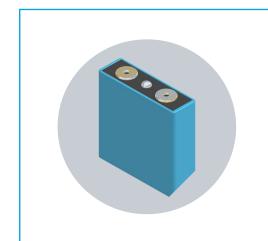


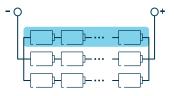






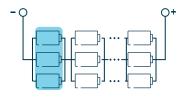
# The cells are stacked to achieve desired voltage and current capacity





**Serial Connection:** In a battery module, the module voltage is equal to the voltage of a single cell multiplied by the number of cells connected in series.

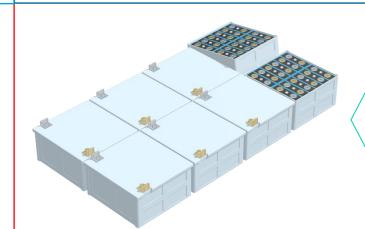
$$V_{module} = N_s \cdot V_{cell}$$



**Parallel Connection:** In a battery module, the current capacity is equal to the current of a single cell multiplied by the number of cells connected in parallel.

$$I_{module} = N_p \cdot I_{cell}$$





The modules connected in series determine the battery pack voltage, and the modules connected in parallel determine the battery pack current capacity

$$V_{PACK} = N_s \cdot V_{module}$$
  
 $I_{PACK} = N_p \cdot I_{module}$ 

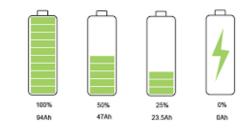






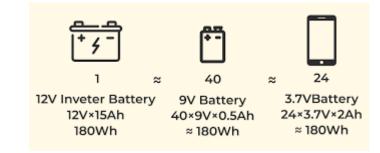
Capacity [Ah]

- + The maximum amount of charge a battery can store and deliver. It Indicates the amount of charge the battery can deliver at a particular discharge rate. It can be measured in Ampere-hour (Ah)
  - + Capacity of 5Ah, means can be discharged for 1 hour at 5A



**Energy [Wh]** 

+ Represents the total power (work) a battery can provide. It can be calculated with the product of capacity and nominal voltage





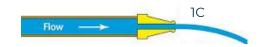




#### C Rate [C]

- + Represents how quick the battery is charged or discharged with a constant current in relation to its maximum capacity. It is expressed as a multiple of the battery's rated capacity
- + For example, in a **5Ah battery**:
  - + 1C rate indicate the battery charge or discharge at 5A, the process will finish in 1 hour.
  - + 2C rate indicate the battery charge or discharge at 10A, will finish in 0.5 hours.
  - + **0.5C** rate indicate the battery charge or discharge at **2.5A**, will finish in **2 hours**.







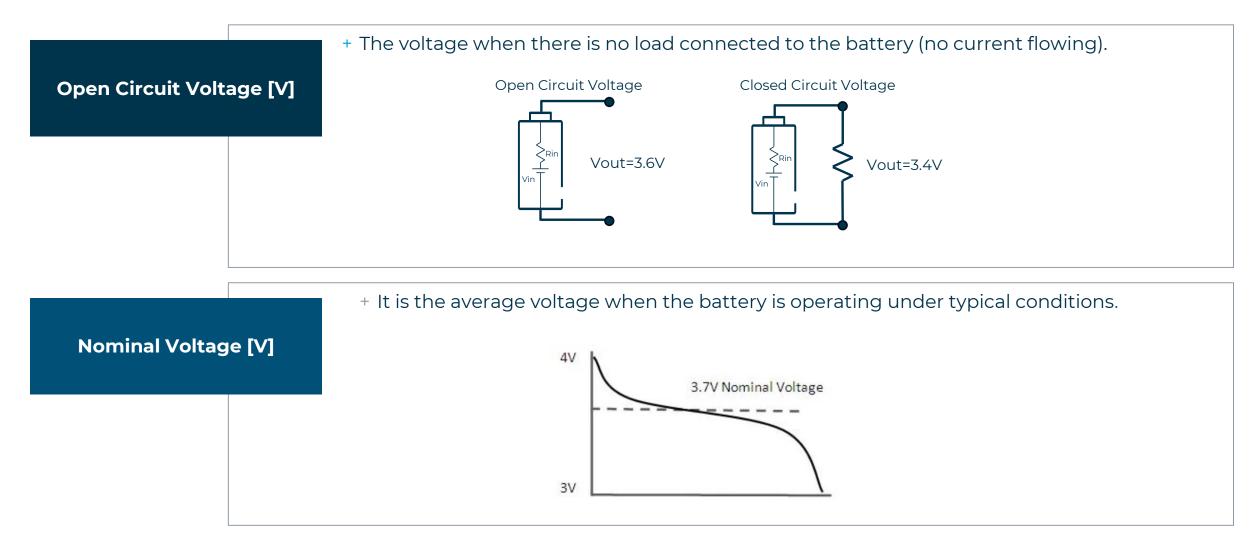
# Energy Density [Wh/L, Wh/kg]

- + The amount of energy can be stored in a determined volume or mass. A higher value of this parameter means smaller ang lighter battery pack.
  - + **Volumetric Energy Density**: Energy stored in a given volume, typically measured in watts-hour per liter (**Wh/L**)
  - + **Gravimetric Energy Density**: Energy stored in a given mass, typically measured in watts-hour per kilogram (**Wh/kg**)











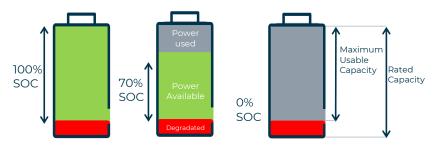




State of Charge (SoC) [%]

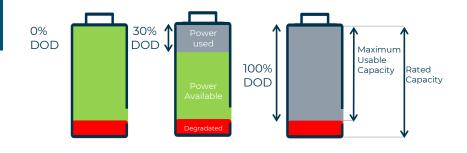
+ Represents the current charge level of energy stored in a battery, it is expressed as the percentage of its capacity.

$$SoC = \frac{Available\ Capacity\ [Ah]}{Maximum\ Capacity\ [Ah]} \times 100\%$$



Depth of Discharge (DoD) [%]

+ Indicates the percentage of the battery that has been discharged relative to its capacity.



$$DoD = 100\% - SoC$$



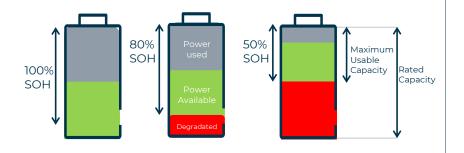




State of Health (SoH)

+ Metric used to compare the condition of the battery compared to its new or ideal state.

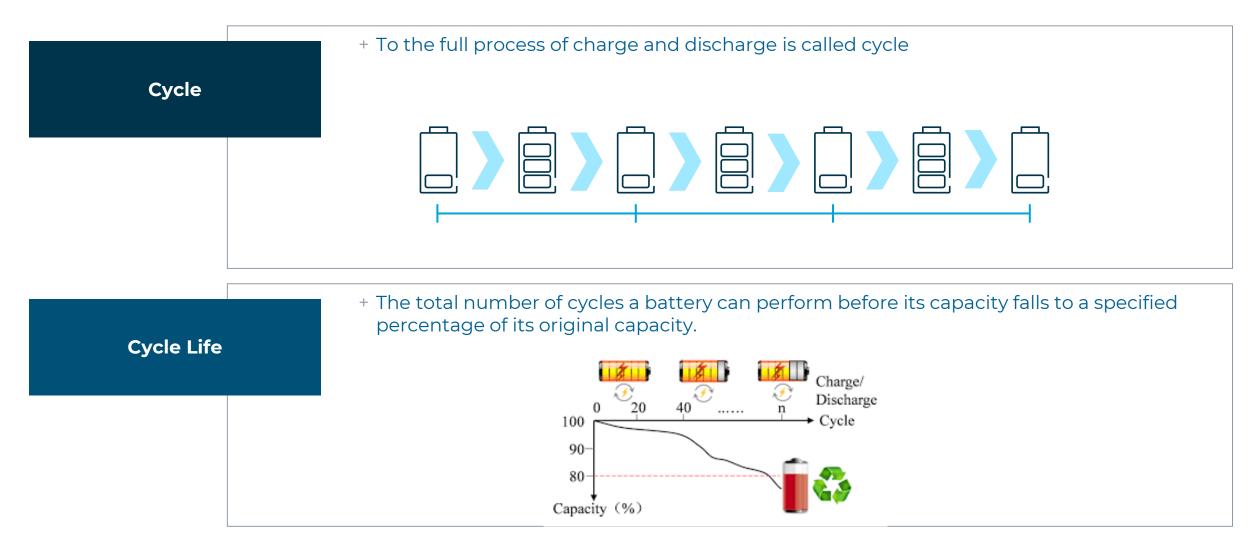
$$SoH = \frac{Maximum\ Capacity\ [Ah]}{Rated\ Capacity\ [Ah]} \times 100\%$$

















## Example - Battery cell Datasheet

+ Capacity = 4.9 Ah

 $+ Energy = 4.9Ah \times 3.6V$ = 17.64 Wh

Item	Specification	
3.1 Standard discharge Capacity	Min 4,900mAh - Charge : 0.5C(2,450mA), 4.2V, 0.02C(98mA) cutoff @ RT - Discharge : 0.2C(980mA), 2.5V cutoff @ RT * 1C = 4,900mA	
3.2 Rated discharge Capacity	Min 4,753mAh - Charge : 0.5C(2,450mA), 4.2V, 0.02C(98mA) cutoff @ RT - Discharge : 1C(4,900mA), 2.5V cutoff @ RT	
3.3 Charging Voltage	4.2V	
3.4 Nominal Voltage	3.6V	
3.5 Charging Method	CC-CV (constant voltage with limited current)	
3.6 Charging Current	Standard charge: 2,450mA	
3.7 Charging Time	Standard charge: 3hours	
3.8 Max. Charge Current	4,900mA (not for cycle life)	
3.9 Max. Discharge Current	9,800mA (for continuous discharge) 14,700mA (not for continuous discharge)	
3.10 Discharge Cut-off Voltage	2.5V	
3.11 Cycle life	Capacity ≥ 3,802mAh @ after 500cycles (80% of the Rated Discharge Capacity @ RT) -Charge: 0.5C(2,450mA), 4.2V,CCCV 0.05C(245mA) cut-off @ F - Discharge: 1C(4,900mA), 2.5V cut-off @ RT	
3.13 Recovery characteristics	Capacity recovery (after the storage) ≥ 4,278mAh (90% of the Rated Discharge Capacity @ RT)  - Charge : 0.5C(2,450mA), 4.2V, 0.02C(98mA) cutoff @ R'  - Storage : 30 days @ 60°C  - Discharge : 1.0C(4,900mA) 2.50V cut-off @ RT	
3.14 Cell Weight	69g max	
3.15 Cell Dimension	Cell height: Max.70.80mm Diameter: Φ Max.20.25mm	



Calculte the Energy Density

Energy Density = 
$$\frac{17.64Wh}{0.069kg}$$
 = 255.65  $\frac{Wh}{kg}$ 

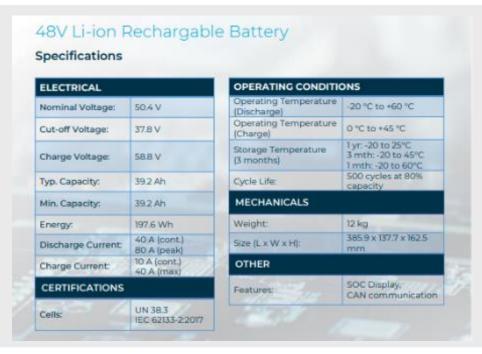






## Example - Battery Pack Datasheet





- +  $Pack \ voltage = 3.6V \times 14$ = 50.4V
- + Pack Capacity =  $4.9Ah \times 8$ = 39.2Ah
- +  $Pack Energy = 39.2Ah \times 50.4V$ = 1.583kWh

Calculte the Energy Density

Energy Density = 
$$\frac{1.583kWh}{12kg}$$
 = 164.64  $\frac{Wh}{kg}$ 







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#### Battery Management Systems (BMS)

#### What is a BMS?

- The Battery Management System (BMS) is a component that monitors, controls and protects the battery pack.
- > It is an essential part of the battery, without its management the battery and specially the user are unprotected from malfunctions of the battery, as over-charging, over-discharging, short-circuits, over-heating, etc.



#### **Key functions and features**

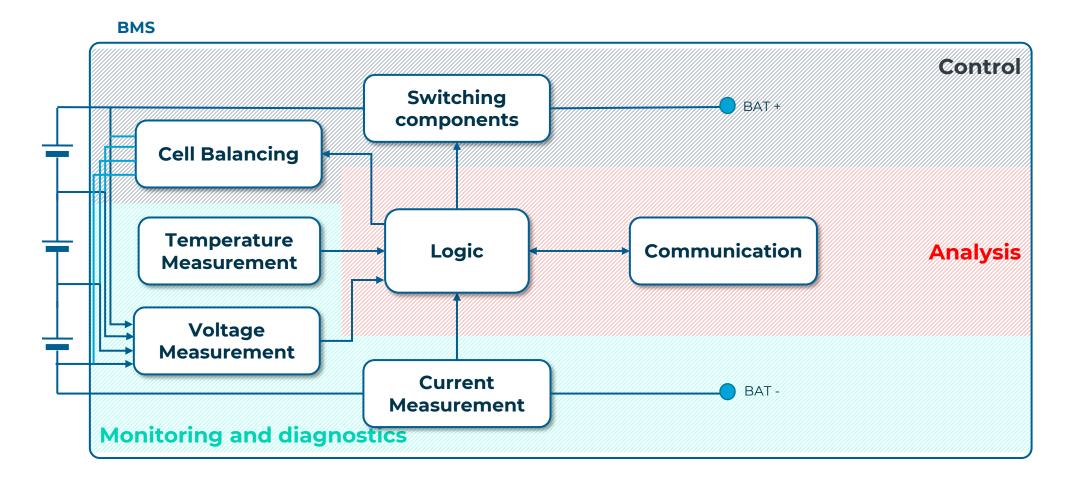
- Battery Monitoring
- Battery Control
- **>** Battery Estimations
- **>** Communications







# BMS is composed by 3 principal parts: Control, Analysis, Monitoring and diagnostics

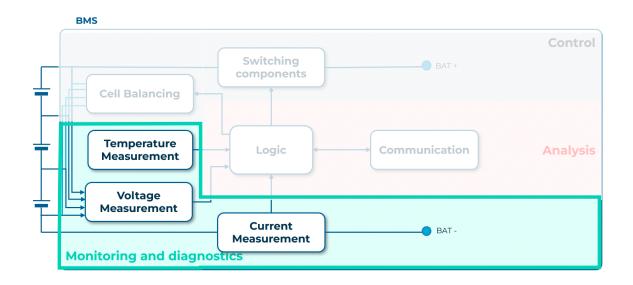








# Monitoring and Diagnostics



#### **Core Features**

- + High-resolution measurement of cell voltage, temperature, and current
- + Integration of sensors for accurate readings.
  - + Pack Voltage and Current
  - + Cells Voltage
  - + Cells Temperature
  - + Insolation
- + Real-time detection of anomalies, including overvoltage undervoltage, and thermal irregularities.

#### **Definition**

- + Continuously measures and collects critical data, such as cell voltage, temperature, and current.
- + Includes all BMS sensors, providing real-time battery state.
- + Identifies deviations from normal operation to detect potential issues.

#### **Remarks**

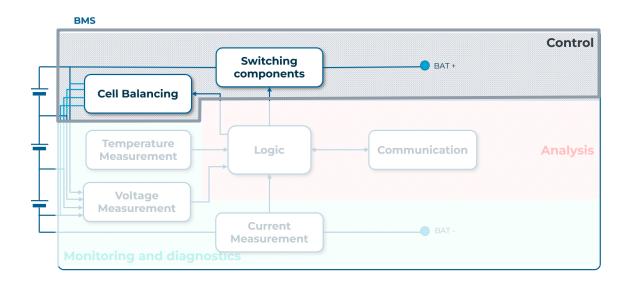
High precision is vital to prevent undetected system faults.







### Control and Protections



#### **Core Features**

- + Dynamic control over power flow to optimize energy usage.
- + Activation of emergency disconnects during unsafe conditions as:
  - + Short circuit
  - + Overload
  - + Overcharging
  - + Overheating
- + Management of cell balancing to maintain uniform SOC across the battery pack.

#### **Definition**

- + Ensures the safe operation of the battery pack by enforcing operational limits and implementing safeguards.
- + Protects against critical risks like overvoltage, undervoltage, overcurrent, and thermal runaway.
- + Maintains the health and performance of the battery pack by controlling energy flow and cell balancing.

#### **Remarks**

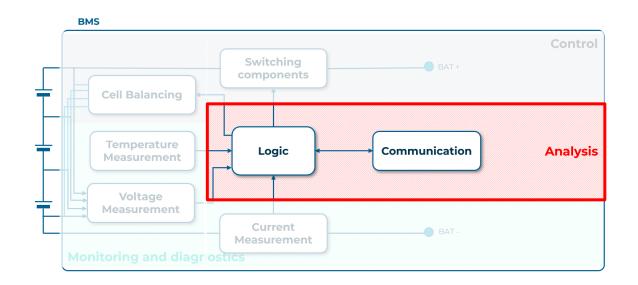
+ Effective protections significantly reduce the risk of catastrophic failures.







# **Analysis**



#### **Core Features**

- + Estimations for battery performance tracking:
  - + State of Health (SoH)
  - + State od Charge (SoC)
- + Communication via industry-standard protocols such as CAN, LIN, and Ethernet.

#### **Definition**

- + Processes collected data to provide information of the battery performance.
- + Estimates critical parameters like SOC and SOH using advanced algorithms.
- + Facilitates seamless integration with external systems through communication protocols..

#### **Remarks**

- Estimation accuracy is key to optimizing battery performance and lifecycle.
- + Robust communication protocols ensure reliability in interconnected systems.







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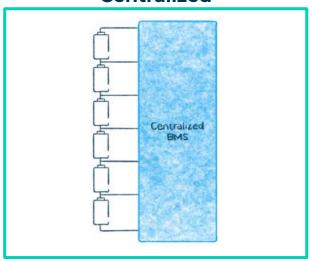




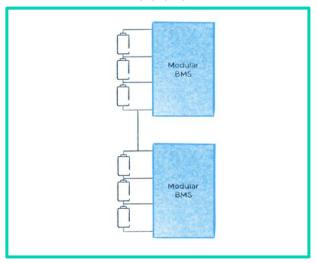
# BMS Topology

#### Depending on the Battery requirements, the BMS can be designed with the following topologies

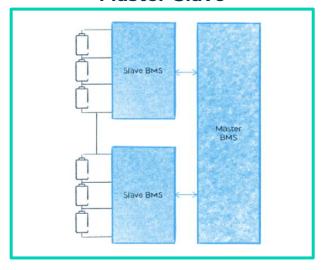
#### Centralized



#### Modular



#### **Master-Slave**









# Centralized Topology

#### Operation

All the battery cells or modules are managed by only one BMS



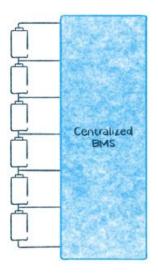
#### **Use Cases**

The most common use of this kind of topology is in smaller and less complex battery systems. In 2-3 wheeler vehicles this option is the most prevalent choice.



#### **Advantages**

- + The design of a system with the control and monitoring in only one central unit simplify and reduce the work time
- + Reduce the time and effort on designing, assembling and maintaining, also reduce the cost of this kind of BMS
- + Communication with only one board ensures higher and synchronous data collection



#### **Disadvantages**

- + Is not a suitable option for larger battery packs or escalable cases
- + Any malfunction in this component could potentially damage the entire battery.
- + Connecting all the sensors of the battery to the centralized BMS create a complex wiring design.







## Modular Topology

#### Operation

The battery pack is divided into multiple modules, and each one is monitored and managed by a different BMS.



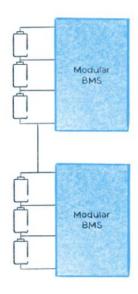
#### **Use Cases**

This topology is especially useful in applications where the battery size can vary over time, like grid energy or backup power systems.



#### **Advantages**

- + Since each module functions independently, expand or reduce the system is made adding or removing moduls.
- + If one of the modules fails, the rest of the system keep working normally
- + Wiring complexity is reduced, each module only needs to connect to a section of the complete battery



#### Disadvantages

- + The implementation can be more challenging because each module works independently while still maintaining coordination between them
- + The cost higher than the centralized BMS, while each module needs its own manufacturing and components
- + The autonomy of cell groups can complicate balancing efforts







## Master-Slave Topology

#### Operation

Adopts a master-slave configuration. Battery cell monitoring is distributed across multiple slave boards, which communicate with the master to manage the entire battery pack.



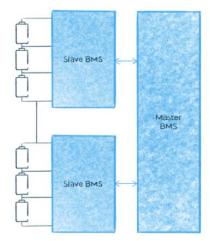
#### **Use Cases**

The distributed topology is most used in large applications, as energy storage systems, aerospace field and electric cars.



#### **Advantages**

- + The system can be scaled by simply adding more slaves BMS.
- + Slave modules handle localized monitoring and control, reducing computational load on the master.
- + The master provide a single point of control and data simplifying decision-making.



#### **Disadvantages**

- + If the master unit fails, the entire system's functionality may be compromised.
- + Communication between the master and slaves requires robust design, adding complexity.
- + The inclusion of multiple slave modules increases the overall system cost compared to simpler topologies.





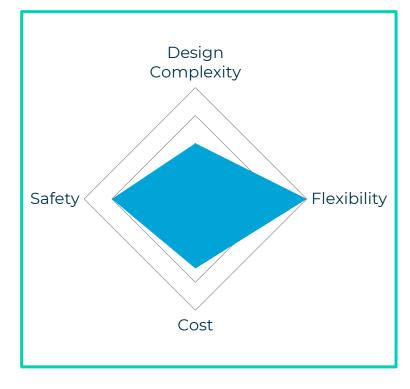


# BMS Topology Comparison

#### Centralized

# Design Complexity Safety Flexibility Cost

#### Modular



#### **Master-Slave**

